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(54) Title: RADIATION-CURABLE AQUEOUS URETHANE (METH)ACRYLATE DISPERSIONS BASED ON ISOCYANATES, (METH)ACRYLATE MONOMERS AND AN INTERNAL EMULSIFIER

(57) Abstract

The invention relates to radiation-curable aqueous urethane (meth)acrylate dispersions based on disocyanates, (meth)acrylate monomers and an internal emulsifier. The internal emulsifier is radiation curable and comprises maleic anhydride units and alkylene oxide units. The dispersion results in coatings being scratch resistant and resistant to chemical substances.

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RADIATION-CURABLE AQUEOUS URETHANE (METH)ACRYLATE DISPERSIONS BASED ON ISOCYANATES, (METH)ACRYLATE MONOMERS AND AN

INTERNAL EMULSIFIER

The invention relates to radiation-curable aqueous urethane (meth)acrylate dispersions based on isocyanates, (meth)acrylate monomers and an internal emulsifier.

Such dispersions are described by Oldring in 'Chemistry and Technology of UV and EB formulation for coatings, inks and paints' (Volume 2, 1991, pp. 185-206).

A drawback of these dispersions is that they result in coatings having an insufficient scratch resistance and having an unsufficient resistance to chemical substances after curing.

The object of the invention is to provide a radiation-curable urethane(meth)acrylate dispersion which results in coatings having good scratch resistance, being resistant to chemical substances and having also other desired properties such as for example a good flow, sandability, gloss and emulsion stability.

The invention is characterised in that the internal emulsifier comprises (1-4) units of maleic anhydride (MA).

Preferably the emulsifier comprises (1-2) units
of maleic anhydride. This emulsifier is radiation-curable.

Preferably, the emulsifier also comprises
alkylene oxide units.

Preferably the alkylene oxide is ethylene oxide. Other suitable alkylene oxides include, for example, propylene oxide and ethylene oxide/propylene oxide block copolymers.

The use of th emulsifier according to the

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invention ensures that the acrylate groups react with on another during the radiation curing of the film while the double bond (resulting from MA) in the emulsifier system also takes part in the reaction. As a result, a hard coating with a good resistance to chemical substances and good scratch resistance is obtained after curing. Moreover this coating can be very well sanded. Other desired properties, such as for example a good hardness, a good flow, and a good gloss are also obtained.

The incorporation of the reactive emulsifier according to the invention also results in an improved emulsion stability.

According to a preferred embodiment of the present invention the emulsifier is represented by formula 15 (I):

20 in which:

a = 5-50, R^1 = (C_1-C_4) alkyl, R^2 = H or CH_3 ,

and

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in which

 R^4 = (C_6-C_{75}) aromatic compound, (C_6-C_{75}) cycloalkyl or (C_6-C_{75}) alkyl and

the aromatic compound, cycloalkyl and alkyl are based on diglycidyl compounds.

The preparation of the emulsifier system

35 according to formula (I) can take place by carrying out,
in a first step, a reaction between an alkoxypolyalkylene
glycol and maleic anhydrid, at temperatures between 100°C

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and 200°C, preferably between 110°C and 130°C, for a period of preferably 1-3 hours. In a second step, the product obtained in the first step is then reacted with a diglycidyl ether, for 1-3 hours, at temperatures of between 100°C and 200°C, preferably between 140°C and 150°C.

Suitable alkoxypolyalkylene glycols include, for example, methoxypolyethylene glycol (MPEG), butoxy-polyethylene glycol and alkoxy copolymers of ethylene glycol and propylene glycol.

Preferably use is made of MPEG. The molar weight is preferably between 500 and 1000.

Suitable diglycidyl ethers include for example the diglycidyl ether of bisphenol A, diglycidyl ether of butane diol, diglycidyl ether of neopentyl glycol and the diglycidyl ether of cyclohexane dimethanol.

Preferably use is made of the diglycidyl ether of bisphenol A.

The molar ratio alkoxypolyalkylene glycol to
20 acid anhydride is usually between 1:0.9 and 1:1.1 and is
preferably between 1:1.0 and 1:1.05. The molar ratio of
the reaction product of alkoxypolyalkylene glycol and
anhydride to diglycidyl ether is usually between 1:0.45
and 1:0.6 and is preferably between 1:0.50 and 1:0.55.

The reactions can be carried out either solventfree or in the presence of solvents such as, for example, toluene, xylene and N-methylpyrrolidone. The reactions are preferably carried out in the absence of solvents.

The two reactions can be carried out either with or without catalysts. Suitable catalysts include, for example, tertiary amines such as for example dimethylbenzylamine and triethylamine, phosphines, phosphonium compounds and tetraalkylammonium halogenides such as for example tetramethylammonium bromide. The catalysts can be used in amounts of between, for example, 0.025 and 0.5 % by weight.

According to a preferr d mbodiment of the

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present invention, in formula (I):

a = 15-20,

 R^1 = methyl,

 $R^2 = H$

5 and

$$R^{3} = -CH_{2}-CH-CH_{2}-O-R^{4}-O-CH_{2}-CH-CH_{2}-OH$$

in which $R^4 = -(C_6H_4)-C_3H_6-(C_6H_4)-$

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According to another preferred embodiment of the present invention the emulsifier is represented by formula (II):

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$$R^6$$
 O O $R^5 - (O-CH_2-CH)_b - O-C-CH=CH-C-O-R^7$ (II)

in which:

b = 5-50,

20 $R^5 = (C_1 - C_4)$ alkyl,

 $R^6 = H \text{ or } CH_3$,

 $R^7 = -CH_2 - CH - CH - R^{\theta}$

OH OH

in which $R^8 = H$ or (C_1-C_4) alkyl.

25 The preparation of the emulsifier system according to formula (II) can take place by carrying out, in a first step, a reaction between an alkoxypolyalkylene glycol and maleic anhydride, at temperatures between 100°C and 200°C, preferably between 110°C and 130°C, for 1-3
30 hours. The reaction product obtained then reacts, in a second step, with an epoxy alcohol, for 1-3 hours, at temperatures between 100°C and 200°C, preferably at temperatures between 130°C and 140°C.

The alkoxypolyalkylene glycol used is preferably methoxypolyethylene glycol (MPEG). The molar weight is preferably between 500 and 1000. Other suitable alkoxypolyalkylene glycols are for example

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butoxypolyalkylene glycol and alkoxy copolymers of ethylene glycol and propylene glycol.

The epoxy alcohol used is preferably glycidol.

The molar ratio of alkoxypolyalkylene glycol to anhydride is usually between 1:0.9 and 1:1.1 and is preferably between 1:1.0 and 1:1.05. The molar ratio of the reaction product of alkoxypolyalkylene glycol and anhydride to epoxy alcohol is usually between 1:0.9 and 1:1.15 and is preferably between 1:1.0 and 1:1.1.

The two reactions can be carried out either with or without catalysts. Suitable catalysts include, for example, tertiary amines such as dimethylbenzylamine and triethylamine, phosphines, phosphonium compounds and tetraalkylammonium halogenides such as tetramethylammonium bromide. The catalysts can be used in amounts of between, for example, 0.025 and 0.5 wt.%.

The reactions can be carried out either solventfree or in the presence of solvents such as for example toluene, xylene or N-methylpyrrolidone. The reactions are preferably carried out in the absence of a solvent.

According to a preferred embodiment of the present invention, in formula (II):

= 15-20,

R⁵ = methyl,

 $25 R^6 = H,$

 $R^{7} = -CH_{2}-CH-CH-R^{8}$ OH OH

in which

 $R^{\theta} = H.$

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Urethane (meth)acrylate oligomers can for example be prepared via the reaction of diisocyanates, (meth)acrylate monomers containing hydroxyl groups and the emulsifier according to the present invention. To obtain certain desired properties diols and polyols may also be added in addition to said components.

The emulsifier precursor according to formula

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(I) or (II), mixed with the desired diols or polyols, may react with isocyanates, for 1-3 hours, at temperatures between 50°C and 100°C, preferably between 75°C and 85°C. The product obtained in the first step may react in a second step with the (meth)acrylate monomer containing hydroxyl groups, for 3 to 5 hours, at a temperature between 50°C and 100°C, preferably between 75°C and 85°C.

The product obtained (having a temperature of, for example, 75°C-85°C) is then dispersed in water having a temperature of between, for example, 20°C and 50°C. During the dispersing any isocyanate groups still present may react with water or diamine, which results in chain growth.

The molar ratio of the total of hydroxyl groups from the acrylate monomers, diols, polyols and emulsifier to the isocyanate groups molar ratio is usually between 0.6 and 1.2 and is preferably between 0.95 and 1.

The reactions described may optionally be carried out in the presence of a solvent; it is however preferable to carry them out in the absence of solvents.

Suitable isocyanates may be both aromatic and aliphatic. Suitable aromatic diisocyanates include, for example, toluene diisocyanate (TDI) and diphenylmethane diisocyanate (MDI). Suitable aliphatic diisocyanates include for example isophoron diisocyanate (IPDI), hexamethylene diisocyanate (HMDI) and oligomers of HMDI and hydrogenated MDI (H12-MDI). Aliphatic diisocyanates having aromatic rings such as, for example, tetramethylene xylene diisocyanate (TMXDI) and xylene diisocyanate (XDI) may also be used.

Preferably IPDI and TDI are used.

Suitable (meth)acrylate monomers containing hydroxyl groups include for example hydroxyethyl (meth)acrylate and hydroxypropyl acrylate.

Preferably hydroxyethyl acrylate is us d.

Various polymers (having a molar w ight of between approximately 500 and 4000) may be used as the

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diol, provided that they have terminal OH groups. Suitable polymers include for example polypropylene glycol, linear polyesters, polycaprolactones, polycarbonates or polytetrahydrofurans. In addition to these polymeric diols use may be made of low-molecular weight polyols. Suitable diols include for example 1,4-butanediol and 1,6-hexanediol. Suitable polyols include for example trimethylolpropane, pentaerythritol, dipentaerithritol and dimethylolpropane.

The alkylene oxide content of the synthesized product is generally between 0.5 and 50.0 wt.%.

The preparation of aqueous urethane acrylate dispersions is described by Oldring in 'Chemistry and Technology of UV and EB formulation for coatings, inks and paints', (Vol. 2, 1991, pp. 73-79) the disclosure of which is incorporated herein by reference.

The radiation curing of the urethane (meth)acrylate dispersion preferably takes place by means of UV or EB curing. Both curing methods are described by S.J. Bett et al. in the article 'UV and EB Curing' in Jocca (1990 (11), pp. 446-452) the disclosure of which is incorporated herein by reference.

The dispersions according to the invention may be used in the preparation of coatings which can be applied to various substrates such as, for example metal, wood, textile, leather, plastic and paper. Other fields of application include for example the ink and adhesive industries.

The emulsifiers according to the invention, in particular the emulsifier according to formula (I), may also be used in combination with other radiation-curable systems such as, for example, dispersions based on epoxy acrylates and polyester acrylates.

The present invention is illustrated with the aid of the following, non-restrictive, examples.

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Example I

Preparation of an emulsifier system according to formula (I)

714 parts by weight of methoxypolyethylene
5 glycol (MPEG) and 98 parts by weight of maleic acid
anhydride (MA) were heated to 120°C in a vessel provided
with a stirrer, under nitrogen. After 3 hours at 120°C (at
an acid number of 71 mg of KOH/g of resin) 188 parts by
weight of diglycidyl ether of bisphenol (Epikote 828TH)
10 were added. After 2 hours at 145°C an acid number of < 5
was reached.

The reaction product had the following constants:

hydroxyl value: 58 mg of KOH/g of resin; 15 acid number: < 5 mg of KOH/g of resin.

Example II

Preparation of an emulsifier system according to formula (II)

20 804 parts by weight of methoxypolyethylene glycol (MPEG) and 111 parts by weight of maleic anhydride (MA) were heated to 120°C under nitrogen in a vessel provided with a stirrer. After 3 hours at 120°C (at an acid number of 71) 85 parts by weight of glycidol were 25 added. After 2 hours at 135°C an acid number of < 5 was reached.

The reaction product had the following constants:

hydroxyl value: 120 mg of KOH/g 30 acid number: < 5 mg of KOH/g.

Example III

Preparation of an urethane acrylate dispersion

155 parts by weight of the emulsifier system
35 according to Example I, 24 parts by weight of
trimethylolpropane (TMP) and 153 parts by weight of
isophoron diisocyanate were mix d and the mixture was

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heated to 90°C. After 3 hours at 90°C, 68 parts of hydroxyethyl acrylate (HEA) were added dropwise to the mixture, which had an NCO content of 8.7%. After 5 hours' reaction at 90°C the NCO content was 1%. Then the product was pumped into a dispersing vessel, into which 600 parts by weight of demineralized water had been introduced. Chain growth subsequently took place. The dispersion was drained at an NCO content of 0%.

10 Example IV

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Preparation of a urethane acrylate dispersion

according to Example II, 21 parts by weight of trimethylolpropane (TMP) and 168 parts by weight of isophoron diisocyanate were mixed and the mixture was heated to 90°C. After 3 hours at 90°C 77 parts of hydroxyethyl acrylate (HEA) were added dropwise to the mixture, which had an NCO content of 9.8%. After 5 hours' reaction at 90°C the NCO content was 1%. Then the product was pumped to a dispersing vessel, into which 600 parts by weight of demineralized water had been introduced. Chain growth subsequently took place. The dispersion was drained at an NCO content of 0%.

25 <u>Example V</u>

UV curing

4% by weight solid substance of a 3.2 gram of 50% solution of 1-hydroxycyclohexylphenylketone (Irgacure 184^{TM}) in ethanol was added to 100 grams of the dispersion according to Example III. This dispersion was applied to a wooden panel with the aid of a $30-\mu\text{m}$ wire coater. This was then dried for 5 min. at 50°C to evaporate the water, after which crosslinking took place as a result of irradiation using a medium-pressure Hg lamp (dose 750 mJ/cm²). Two layers were applied, the first of which was sanded with sandpaper after application.

The characteristics of th coating are summarized in Table 1.

TABLE 1

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particle size:	80 nm
weight per unsaturation:	510
König hardness (1)	125
scratch resistance (2)	1
resistance to ethanol (3)	1
resistance to xylene (3)	0
resistance to coffee (3)	0

(1) determined by DIN 53157

15 (2) determined visually using a wire-brush:

0 = very good

1 = good

4 = poor

2 = satisfactory

3 = moderate

20

(3) determined according to DIN 68 861-1A:

0 = very good

1 = good

2 = satisfactory

25 3 = moderate

4 = poor

These examples show that the internal emulsifier according to the present invention results in a good scratch resistance and a good resistance to chemical substances of the radiation-cured coating.

CLAIMS

- Radiation-curable aqueous urethane (meth)acrylate dispersion based on isocyanates, (meth)acrylate monomers and an internal emulsifier, characterised in that the internal emulsifier comprises (1-4) units of maleic anhydride.
 - 2. Dispersion according to claim 1, characterised in that the emulsifier comprises alkylene oxide units.
- 10 3. Dispersion according to any one of claims 1-2, characterised in that the emulsifier is represented by formula (I):

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in which:

a = 5-50,

 $R^1 = (C_1 - C_4) \text{ alkyl},$

 $R^2 = H \text{ or } CH_3$

20 and

$$R^{3} = -CH_{2} - CH - CH - O - R^{4} - O - CH_{2} - CH - CH_{2} - CH_{2} - CH - CH_{2} - C$$

in which

25 $R^4 = (C_6 - C_{75})$ aromatic compound, $(C_6 - C_{75})$ cycloalkyl or $(C_6 - C_{75})$ alkyl and

where the aromatic compound, cycloalkyl and alkyl are based on diglycidyl compounds.

30 4. Dispersion according to claim 3, characterised in that in formula (I):

a = 15-20,

R¹ = methyl,

 $R^2 = H$

35 and

$$R^3 = -CH_2 - CH - CH_2 - O - R^4 - O - CH_2 - CH - CH_2 - OH$$
 OH

in which $R_4 = -(C_6H_4)-C_3H_6-(C_6H_4)-.$

- Use of the emulsifier according to any one of claims
 1-4 in the preparation of radiation-curable systems.
 - 6. Use of the emulsifier according to formula (I) in the preparation of radiation-curable systems.
 - 7. Use of a dispersion according to any one of claims 1-4 in the preparation of coatings.
- 10 8. Coating based on a dispersion according to any one of claims 1-4.
 - 9. Entirely or partly coated substrate, characterised in that the substrate is coated with a coating according to claim 8.

INTERNATIONAL SEARCH REPORT

Intera aal Application No

•			PCT/NL 94/00113		
A. CLASSI	IFICATION OF SUBJECT MATTER C09D175/16 C08G18/67				
	3332,0,23				
According to	to International Patent Classification (IPC) or to both national class	sification and IPC	<u> </u>		
B. FIELDS	SEARCHED				
Minimum de IPC 5	locumentation searched (classification system followed by classific CO9D CO8G	auon symbols)	•		
Documentat	tion searched other than minimum documentation to the extent that	nt such documents are includ	led in the fields searched		
Electronic d	lata base consulted during the international search (name of data b	ase and, where practical, see	arch terms used)		
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